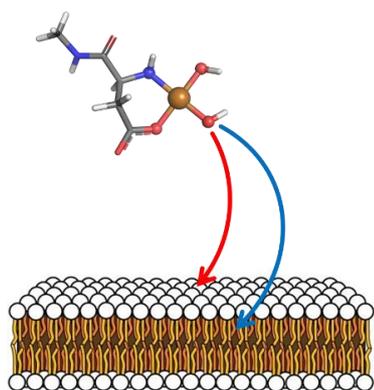


Membrane damages induced by Cu(II)-A β ·OH radical species. OH propagation toward polar head groups and lipid tails of membrane phospholipid bilayers at DFT level.

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The Oxidative stress [1] appears to be one of the major determinants of the pathogenesis and progression of Alzheimer's disease (AD). According to the oxidative stress hypothesis in AD, the interactions of redox active Cu(II) ions with A β peptide is linked to production of toxic reactive oxygen species (ROS). While the Cu(II)-A β ascorbate assisted OH radical production is well characterized, less is known about the following OH propagation processes.



The product that comes from the O₂ ascorbate-assisted reduction cycle is the Cu(II)-A β ·OH species in which one OH radical is coordinated to Cu(II). Besides intramolecular A β oxidative processes [4], OH radical can propagate oxidizing polar head groups and lipid tails of phospholipid bilayers in brain membranes.

We investigate at Density Functional Theory level [2-4] the OH propagation of Cu(II)-A β ·OH model coordinations exploring 1) Cu(II)-A β ·OH/phosphocholine polar head non-bonding interaction; 2) breaking mechanism of the -CH₂-N(CH₃)₃⁺ C-N and glycerol C-O bonds in phosphocholine upon impact of an Cu(II)-A β ·OH radical; 3) The lipid peroxidation initiation step considering a selection of fatty acids singled out among the most abundant in human brain membranes. The results obtained can be useful to elucidate some aspects of the binding and the reactivity of the Cu(II)-A β ·OH in connection with oxidative stress and copper dyshomeostasis in AD.

References

- [1] C. Cheignon, M. Tomas, D. Bonnefont-Rousselot, P. Faller, C. Hureau, F. Collina,, Redox Biology, (2018) 14, 450–464; [2] T. Prosdocimi, L. De Gioia, G. Zampella, L. Bertini. J Biol. Inorg. Chem, (2016), 21, 197–212; [3] F. Arrigoni, T. Prosdocimi, L. Mollica, L. De Gioia, G. Zampella, L. Bertini, Metallomics, (2018) 10, 1618-1630; [4] F. Arrigoni, F. Rizza, R. Tisi, L. De Gioia, G. Zampella, L. Bertini, Metallomics (2020) 12, 1765-1780