

Molybdenum Complexes as Catalytic Cyanide Antagonists: Bicompatibility, Intracellular Distribution and Mechanism.

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Cyanide detoxification *in vivo* is most often accomplished with Co(II) based chelators such as [Co(EDTA)] or Hydroxocobalamin [1]. Enzyme catalyzed conversion of cyanide to thiocyanate by the metal-free rhodanase enzyme is well documented [2]. A few examples are known of cyanide forming thiocyanate when reacting with sulfur bound to molybdenum enzymes [3, 4]. A few examples are known of cyanide forming thiocyanate in a reaction of molybdenum complexes with cyanide [5, 6]. The reactivity of a sulfur abstraction reaction by cyanide with Mo-disulfide moiety to form thiocyanate was studied and expanded to catalytic conditions [7]. High conversion of cyanide to thiocyanate was achieved in a time period of 20 minutes. Catalytic cycle is proposed where initial step requires a sulfur abstraction reaction. Cytotoxicity of a series of complexes was studied as well as *in vivo* toxicity via a dose escalation study [8]. A preliminary pharmacokinetic profile of a selected compound was obtained using ICP-OES, and survival studied at lethal HCN dose in a mouse model [9].

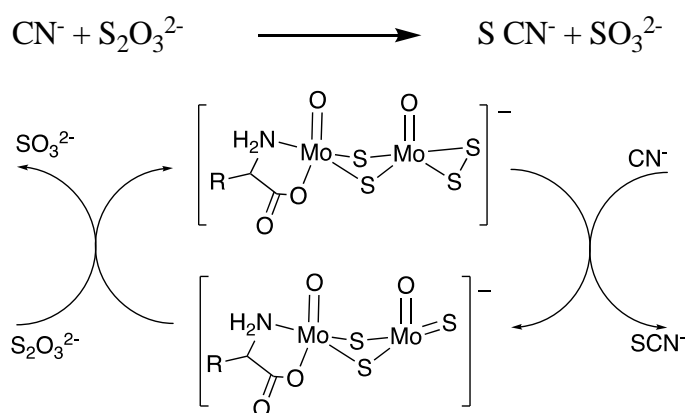


Figure: Reaction of cyanide and thiosulfate catalyzed by $[(\text{RC}_2\text{O}_2\text{NH}_2)\text{Mo}_2\text{O}_2(\mu\text{-S})_2(\text{S}_2)]^-$.

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